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State 7: $T_7 = 400^\circ\text{C}$, $p_7 = 8 \text{ MPa}$

From steam table or CATT2 program

(Computer Aided Thermodynamic Tables 2)

$$h_7 = 3138.3 \text{ kJ/kg}, s_7 = 6.3633 \text{ kJ/kg}\cdot\text{K}$$

Isentropic expansion to state 8, $p_8 = 8 \text{ kPa}$

$$\Rightarrow h_{8,\text{isen}} = 1989.9 \text{ kJ/kg}$$

$$h_8 = h_7 + 0.90 \cdot (h_{8,\text{isen}} - h_7) = 3138.3 + 0.90 \cdot (1989.9 - 3138.3)$$

$$h_8 = 2104.7 \text{ kJ/kg}$$

State 9: Saturated liquid at 8 kPa:

$$h_9 = 173.86 \text{ kJ/kg}, s_9 = 0.59254 \text{ kJ/kg}\cdot\text{K}$$

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